



## Programme Specification

A statement of the knowledge, understanding and skills that underpin a taught programme of study leading to an award from  
The University of Sheffield

1	<b>Programme Title</b>	Foundation Year – Chemistry
2	<b>Programme Code</b>	CHMU99
3	<b>JACS Code</b>	F102
4	<b>Level of Study</b>	Undergraduate – level 0
5a	<b>Final Qualification</b>	None – progression to level 1
5b	<b>QAA FHEQ Level</b>	Level 3
6	<b>Intermediate Qualification(s)</b>	None
7	<b>Teaching Institution</b> (if not Sheffield)	Not applicable
8	<b>Faculty</b>	Engineering
9	<b>Co-ordinating Department</b>	Interdisciplinary Programmes Engineering
10	<b>Other Department(s) involved in teaching the programme</b>	Applied Mathematics, Chemical Engineering Physics & Astronomy, Chemistry
11	<b>Mode(s) of Attendance</b>	Full-time
12	<b>Duration of the Programme</b>	1 year
13	<b>Accrediting Professional or Statutory Body</b>	Not applicable
14	<b>Date of production/revision</b>	September 2021

### 15. Background to the programme and subject area

The Foundation year is open to potential students in Maths, Physics and Chemistry. The content of the course is currently 33.3% Mathematics, 33.3% Physics, and 33.3% Chemistry. Students have to state at application stage which Undergraduate course they will take in 1<sup>st</sup> year if they pass Foundation year. Some students have changed course but this is dependent on the receiving department having capacity.

The main aim of the year is to bring the students academically up to the A level entry requirements, for their department, in maths and sciences. These programmes are only suitable as progression routes onto an undergraduate course at this university.

Further information about the programmes may be found on the internet at <http://www.sheffield.ac.uk/sefy>

### 16. Programme Aims

The aims of the programmes are to:

1. enable students to develop a thorough knowledge and understanding of the elements of maths, physics, chemistry as required of an equivalent A level entry student.
2. provide students with an educational base which will in part satisfy the ability to progress to an undergraduate degree in science.
3. give students the opportunity to study particular aspects of science in depth, according to their interests.
4. encourage in students independence of thought and a critical approach to the interpretation of experimental evidence and to the evaluation of existing information.

5. foster in students the necessary skills needed to be a successful undergraduate.

## 17. Programme learning outcomes

**Knowledge and understanding:** By completion of the year, students will have knowledge and understanding of the:

<b>K1</b>	Mathematics comparable to A-level.
<b>K2</b>	Physics & Chemistry comparable to A-level.
<b>K3</b>	Understand the process of scientific investigation through practical work

### Skills and other attributes

**Intellectual skills:** By completion, students will be able to:

<b>I1</b>	Select the appropriate formula and be able to solve a range of mathematical problems.
<b>I2</b>	Have a basic understanding of dynamics, materials and energy Physics.
<b>I3</b>	Have a basic understanding of organic, inorganic and physical Chemistry
<b>I4</b>	Interpret the results of experimental investigations.

**Practical skills:** By completion, students will be able to:

<b>P1</b>	Demonstrate and describe ethical, safe and skilful practical techniques.
<b>P2</b>	Process and select appropriate qualitative and quantitative methods.
<b>P3</b>	Make, record and communicate reliable and valid observations.
<b>P4</b>	Make measurements with appropriate precision and accuracy.
<b>P5</b>	Analyse, interpret, explain and evaluate the methodology, results and impact of their own and others experimental and investigative activities in a variety of ways.

**Transferable skills:** By completion students will have experience of:

<b>T1</b>	Writing reports in a style appropriate for the audience.
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## 18. Teaching, learning and assessment

**Development of the learning outcomes is promoted through the following teaching and learning methods:**

- **Lectures:** used to transmit information, explain theories and concepts, and illustrate methods of analysis or design. For most lecture courses tutorial sheets are provided to enable students to develop their understanding during private study.
- **Practical classes:** working in groups of two or three, students undertake laboratory experiments to gain practical skills. Labs will have guidance sheets explaining the experiment and aims, and will require students to collect data, graph, calculate parameters and write valid conclusions.
- **Personal tutorials:** run for small groups of six or less to discuss both technical and transferable skill based material. Students are encouraged to take an active part in discussions. These tutorials will also facilitate transfer of specific technical knowledge regarding home department activities, required for the group and individual projects.
- **Problem classes:** run for the whole class to help students to resolve difficulties as they work through the problem sheets.

Opportunities to demonstrate achievement of the learning outcomes are provided through the following assessment methods:

- **Written examinations** - examinations of up to three hours duration.
- **Coursework submissions** - these include formal laboratory reports, group report presentation and poster, individual essay and tutorial assignments.

The main teaching, learning and assessment methods adopted for each learning outcome are shown below. In most cases a combination of methods is used.

Learning Outcome (in abbreviated form – see section 17 for the full text)	Teaching / Learning					Assessment			
	Lectures	Practical classes	Coursework	Tutorials / example	Individual design	Written	Coursework	Oral presentations	Individual project
<b>K1</b> Maths	☑	☑	☑	☑		☑	☑		
<b>K2</b> Physics & Chemistry	☑	☑	☑	☑		☑	☑		
<b>K3</b> Scientific Investigation		☑		☑			☑		
<b>I1</b> Selecting mathematical formula and use		☑	☑	☑		☑	☑		
<b>I2</b> Knowledge – physics		☑	☑	☑		☑	☑		
<b>I3</b> Knowledge – chemistry		X	X	X		X	X		
<b>I4</b> Understanding experimental outcomes		☑	☑	☑			☑		
<b>P1</b> Safe working practices		☑	☑	☑			☑		
<b>P2</b> Qualitative and Quantitative methods		☑	☑	☑			☑		
<b>P3</b> Accurately record data		☑	☑	☑			☑		
<b>P4</b> Use a range of measuring equipment		☑	☑	☑			☑		
<b>P5</b> Analyse practical outcomes		☑	☑	☑			☑		
<b>T1</b> Presentation skills	☑		☑	☑	☑		☑	☑	☑

Proportions of types of assessment by level can be found on the UniStats website: <http://discoveruni.gov.uk/>

## 19. Reference points

**The learning outcomes have been developed to reflect the following points of reference:**

Subject Benchmark Statements

<https://www.qaa.ac.uk/quality-code/subject-benchmark-statements>

Framework for Higher Education Qualifications (2014)

<https://www.qaa.ac.uk/docs/qaa/quality-code/qualifications-frameworks.pdf>

University Strategic Plan

<https://www.sheffield.ac.uk/vision>

Learning and Teaching Strategy (2016-21)

[https://www.sheffield.ac.uk/polopoly\\_fs/1.661828!/file/FinalStrategy.pdf](https://www.sheffield.ac.uk/polopoly_fs/1.661828!/file/FinalStrategy.pdf)

UK-SPEC, Engineering Council, 2020

<https://www.engc.org.uk/standards-guidance/standards/uk-spec/>

## 20. Programme structure and regulations

The structure of these programmes is modular and worth a total of 120 credits. The student enrolls on a Foundation year in MChem Chemistry. The credit structure is appropriate for going on to study MChem or BSc Chemistry. It is possible for a student to change their choice of degree pathway, but only with the agreement of the admissions tutor in the receiving department, and is in no way guaranteed. Successful completion of the 120 credits at 60% or above guarantees the student a place at level 1 in their original department which they applied for only.

A student's initial application in no way constrains this choice, but changes between and within BSc and MChem programmes, and vice versa, are not generally permitted after the beginning of Year 3. Students must satisfy the appropriate progression criteria in order to proceed to Year 3 of an MChem programme; those who do not will be required to join the third year of the BSc programme for this course.

Detailed information about the structure of programmes, regulations concerning assessment and progression and descriptions of individual modules are published in the University Calendar available on-line at <http://www.sheffield.ac.uk/calendar/regs>.

## 21. Student development over the course of study

**On successful completion of the programme (>=60%)** - Students will be automatically accepted onto the original degree course applied for on an undergraduate level 1 programme. Transfer to any other degree course is only possible with the agreement of the receiving department and is not guaranteed.

## 22. Criteria for admission to the programme

Detailed information regarding admission to the programme is available at <http://www.shf.ac.uk/prospective/>

### 23. Additional information

The foundation course is run in collaboration with the Faculty of Science and lectures are jointly attended by students on Foundation Year Physics, Biosciences and Engineering.

Further information about all the programmes and the departments can be found on-line at <http://www.shef.ac.uk/sefy>

This specification represents a concise statement about the main features of the programme and should be considered alongside other sources of information provided by the teaching department(s) and the University. In addition to programme specific information, further information about studying at The University of Sheffield can be accessed via our Student Services web site at <http://www.shef.ac.uk/ssid>.